B.Sc. DEGREE EXAMINATION – JUNE, 2019.

First Year

Chemistry

GENERAL CHEMISTRY — I

Time: 3 hours Maximum marks: 75

PART A —
$$(3 \times 5 = 15 \text{ marks})$$

Answer any THREE questions.

1. Give IUPAC name for the following organic compounds.

(a)
$$CH_3 - CH = CH - CH_2 - CH_2 - CH_3$$

$$\mbox{(b)} \qquad \mbox{H_2N-CH}_2-\mbox{CH}_2-\mbox{CH}_2-\mbox{CH}_3$$

(c)
$$\begin{array}{cc} \operatorname{CH}_3 - \operatorname{CH}_2 - \operatorname{CH}_2 - \operatorname{CH} - \operatorname{CH}_2 - \operatorname{CH}_3 \\ | \\ \operatorname{OH} \end{array}$$

(d)
$$CH_3 - CH_2 - CH - CH_3$$

 CHO

(e)
$$CH_3 - C - CH_2 - CH_3$$

- 2. (a) Write the structure of the following compounds:
 - (i) 3-aminopropan-l-ol
 - (ii) 2-methylpyridine
 - (b) Write short notes on inductive effect.
- 3. How the elements are classified based on the electronic configuration?
- 4. Explain hydrogen bonding with suitable examples.
- 5. Write short notes on viscosity and surface tension.

PART B —
$$(4 \times 15 = 60 \text{ marks})$$

Answer any FOUR questions.

- 6. (a) How the organic compounds are classified based on the C and H atoms?
 - (b) State and explain the IUPAC rules for naming aliphatic compound containing alcohols and amines.
- 7. Write short notes on the followings:
 - (a) resonance effect
 - (b) hyperconjugation
 - (c) steric effect.

2 **UG-350**

- 8. Give an account for the followings:
 - (a) Atomic radii
 - (b) Ionic radii
 - (c) Bond length.
- 9. (a) Write the properties of ionic compounds.
 - (b) Explain octet rule with suitable examples.
- 10. Write a short note on Liquid crystals and its applications.

3

UG-350